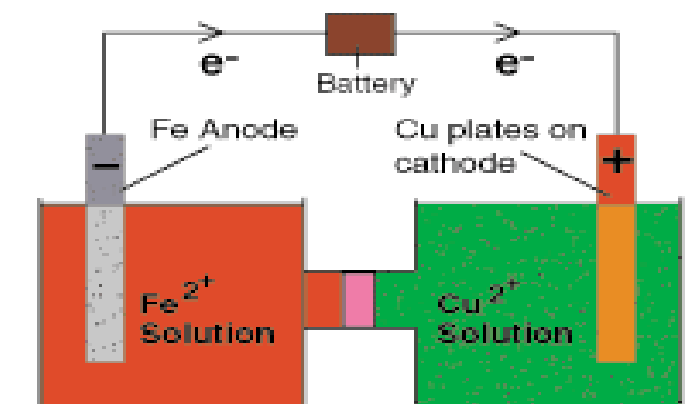


### Galvanic Cell

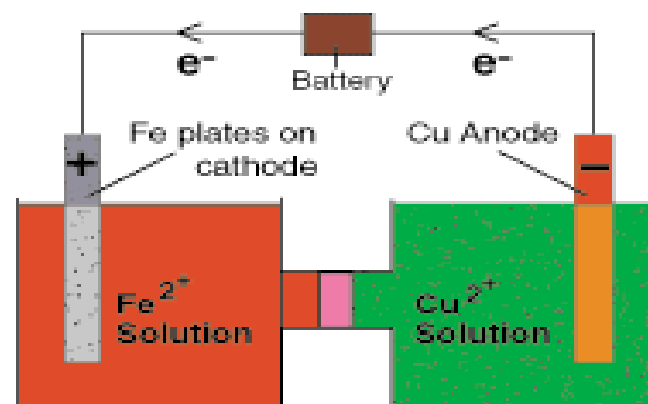


$$E_{\text{cell}}^{\circ} = +0.44 \text{ V}$$

$$E_{\text{cell}}^{\circ} = +0.34 \text{ V}$$

$$E_{\text{cell}}^{\circ} = +0.78 \text{ V}$$

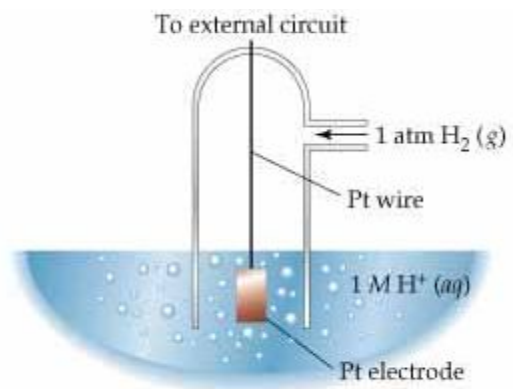
### Electrolytic Cell



$$E_{\text{cell}}^{\circ} = -0.44 \text{ V}$$

$$E_{\text{cell}}^{\circ} = -0.34 \text{ V}$$

$$E_{\text{cell}}^{\circ} = -0.78 \text{ V}$$



(a)

